

ROWAN UNIVERSITY CURRICULUM PROPOSAL

PROPOSAL TITLE: ENGINEERING THERMODYNAMICS II

916-312

CHECK APPROPRIATE: UNDERGRADUATE GRADUATE 2 SEMESTER HOURS

SPONSOR(S): ANTHONY J. MACCHESE

DEPARTMENT/TELEPHONE # MECHANICAL ENGINEERING x 4627

CHECK ONE: COURSE MINOR PROGRAM CONCENTRATION SPECIALIZATION
 ACHIEVEMENT CERTIFICATE CERTIFICATION PROGRAM MAJOR PROGRAM

Step #1 (Department)

Approved (Date) 10/15/97

Not Approved (Date)

[Signature]
Dept. Curriculum Chr.

10/15/97
Reviewed (Date)

[Signature]
Dept. Chr.

Step #2 (Receipt)

SCC# 97-98-68

10-21-97
Date Received Senate

[Signature]
Senate Curriculum Chr.

Step #3 (School)

Reviewed Date: 10/20/97

Recommend to Approved

Recommend NOT to Approve

Forward for Open Hearing:

WITHOUT Reservations

WITH Reservations:
Comments:

[Signature]
School Committee Chr.

Step #4 (Academic Dean): Recommended NOT Recommended Conditionally Recommended (See Comments)

Comments:

Dean Signature/Date

[Signature] 10/20/97

Step #5 (Senate Curriculum Committee): Open Hearing Date: 11-4-97 Approved by Curriculum Committee Date 11-4-97

Returned to Sponsor(s) for the following reason:

Step #6 (Senate) Date announced/voted on at Senate 11-25-97 If voted on: Approved NOT Approved

Forwarded to Executive Vice President/Provost 11-25-97

Senate Curriculum Committee chair Signature/Date: _____

Step #7 (Executive Vice President/Provost): Date Received Jan 28 1998

Approved

NOT Approved If no, reasons are as follows:

Student Credit Hours _____

Faculty Load Hours _____

Equalized Credit Hours _____

Official Copy & Approval Sheet Filed (Date) 1/26/98

Executive Vice President/Provost Signature (J. Mattson)

Registrar

Date Approved Course Description Received 28 Jan 98

Hegis Taxonomy and Course Number Assigned 6410-312

Date/Signature of Registrar B. J. Kelsey

Notification Forward:

_____ Senate Curriculum Committee Chairperson

_____ Department Chairpersons

_____ Academic Dean(s)

_____ Registrar

_____ Sponsor(s)

Course Proposal

1. Details:

a) Course Title:	Engineering Thermodynamics II (0910-312)
b) Sponsor:	Dr. Anthony J. Marchese, Department of Mechanical Engineering, College of Engineering
c) Credit Hours:	2 credit hours
d) Course Level:	Junior
e) Curricular Effect:	Required course for mechanical engineering majors; An elective for other engineering majors
f) Prerequisites:	Engineering Thermodynamics I or Chemical Process Principles II
g) Suggested Time/ Scale of Implementation	Fall 1998 One section
h) Resources:	Faculty is in place to teach the course within the Dept. of Mechanical Engineering. A new 1130 square foot thermodynamics and engine laboratory will be dedicated to this course. Laboratory equipment has already been ordered to support this course. Library resources are in place. Computer hardware resources are available in the Engineering Building to support this course. No new software resources are required.

2. Rationale:

The proposed course is part of the Engineering Curriculum Proposal approved by the College Senate in December 1994. The proposed course is consistent with the establishment of the School of Engineering approved by the Board of Trustees in February 1995.

The course directly follows Engineering Thermodynamics I and consists of applying the fundamental theory to the solution of gas power systems and refrigeration systems. In addition to the applications, more advanced thermodynamic principles are covered, such as thermodynamic relations of simple compressible substances, non-reacting ideal gas mixtures, psychometrics and reacting mixtures. An up-to-date introduction to availability (exergy) analysis is also presented.

The sequence of Engineering Thermodynamics I and II, which is required for all mechanical engineering majors, is the foundation for all following courses in the thermal/fluid science area.

3. Essence of the Course:

a) Objectives:

Having provided students with the fundamental background and analytical tools of thermodynamics in Engineering Thermodynamics I, the main objective of Engineering Thermodynamics II is to provide mechanical engineering students with practical applications that use the principles. In addition, more advanced thermodynamic topics are covered, such as thermodynamic relations of simple compressible substances, non-reacting ideal gas mixtures, psychometrics, reacting mixtures, phase equilibrium and chemical equilibrium.

Upon completion of the course, students will be able to:

1. Solve problems involving mobile gas power plants (Otto cycle, Diesel cycle, Brayton cycle, etc.),
2. Determine the performance and efficiency of refrigeration systems (vapor compression refrigeration, absorption cycles, heat pumps),
3. Determine the composition of a chemical reacting system at equilibrium and calculate the adiabatic flame temperature.
4. Perform availability (exergy) analyses on open and closed systems
5. Calculate vapor pressures for two-phase systems.
6. Use the Maxwell relations to construct tables of thermodynamic properties.
7. Design air-conditioning systems using the psychrometric principles.

b) Topical Outline:

The topical outline of the course may vary to some extent depending on the interests of the instructor and the students, and the advances in engineering technology. The topics to be covered will include the following:

Gas power systems

- Otto and diesel cycles
- Brayton cycle
- Gas turbine engines
- Ericsson and Stirling cycles
- One-dimensional steady flow in supersonic nozzles

Refrigeration and heat pump systems

- Vapor-compression refrigeration
- Refrigerant properties
- Absorption systems
- Heat pumps
- Gas refrigeration systems

Availability (exergy) analysis

Availability and flow availability

Availability balances for closed and open systems

Second law efficiency and thermoeconomics

Thermodynamic relations for simple compressible substances

Equations of state

Maxwell relations

Constructing tables of thermodynamic properties

Generalized charts for enthalpy and entropy

Non-reacting ideal gas mixtures

p-v-T relationships for ideal gas mixtures

Mixing of ideal gases

Psychrometric principles

Conservation of mass and energy for psychrometric systems

Adiabatic-saturation and wet-bulb temperature

Psychrometric charts and applications

Chemically reacting systems

Combustion processes

Conservation of energy for reacting systems

Adiabatic flame temperature

Chemical and phase equilibrium

Equation of reaction equilibrium

Calculation of equilibrium compositions

Phase equilibrium

Suggested Textbook: Fundamentals of Engineering Thermodynamics, Moran and Shapiro

c) Evaluation and Grading Procedure of Students:

Student grades will be determined on the basis of examinations, homework and/or projects, laboratory projects and reports.

d) Course Evaluation:

The proposed course will be evaluated on the basis of student evaluations and curriculum review by appropriate faculty.

4. Results of Consultations:

The proposed course is part of the Engineering Curriculum Proposal approved by the Faculty Senate in December 1994. Consultations were submitted with original proposal as specified by the Curriculum Committee.

Catalog Description:

Engineering Thermodynamics II (0910-312)

(Prerequisites: *Engineering Thermodynamics I* or *Chemical Process Principles II*)

This course emphasizes application of the first and second laws of thermodynamics to a variety of systems, including gas power systems, refrigeration systems and combustion systems. In addition, advanced thermodynamics topics are covered, including Maxwell relations, exergy analysis, chemical equilibrium and phase equilibrium.